Influence of indium and hydrogen co-doping on optical and electrical properties of zinc oxide thin films deposited by DC magnetron sputtering

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ABSTRACT

Introduction: ZnO-based thin films, known as potential transparent-conducting oxides (TCO), have still attracted much attention in applications for good-performance electrodes and inner layers in solar cells. Recently, the research tendency has focused on improving carrier mobility rather than carrier concentration to enhance performance and response speed of TCO thin films. In this work, Indium, and Hydrogen co-doped ZnO (HIZO) thin films were deposited by using DC magnetron sputtering technique in hydrogen-plasma atmosphere. **Methods**: Indium-doped ZnO ceramics were used as sputtering targets, in which, Indium content varied from 0.07 to 1.0 at.%. The electrical, optical, structural and surface morphological properties of the as-deposited films were investigated by using Hall effect-based measurement, UV-Vis spectra, X-ray diffraction (XRD) and field-emission scanning electron microscopy (FE-SEM), respectively. **Results**: As a result, the HIZO films sputtered from the 0.1 at.% In-doped ZnO target and at $H_2/(H_2+Ar)$ ratio of 3.5% exhibit high electron mobility (47 cm²/Vs), the lowest resistivity ($4.9 \times 10^{-4} \Omega$.cm) and sheet resistance (4.7Ω /sq.), simultaneously, high average transmittance (>80%) in the visible – near IR spectrum regions. **Conclusion**: Based on these results, the HIZO films are considered as potential TCO thin films that can be well-used as transparent electrodes in solar cells.

Key words: indium and hydrogen co-doped ZnO, magnetron sputtering, TCO thin films, transparent electrodes

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INTRODUCTION

Transparent conducting oxide (TCO) thin films play an essential role in optoelectronic devices. Until now, Sn-doped In₂O₃ (ITO) has still been the best TCO with preeminent electrical and optical properties, which used as transparent electrodes. Because of the scarcity of indium, however, it is essential to explore new alternative TCO materials alternative for ITO, which has attracted much attraction of researchers around the world. Based on the potential properties of ZnO material, the IIIA-group elements (such as Al, Ga, In) doping into ZnO thin films can improve the conductivity owing to the increase of carrier concentration¹⁻³. On the other hand, the increased carrier concentration often reduces the optical transmittance significantly, especially in the near-IR and IR spectrum regions, due to the free-carrier absorption effect^{4,5}. To solve this problem, increasing carrier mobility is expected to be more effective than carrier concentration.

Hydrogen (H) is known as a dopant which can improve carrier mobility of ZnO films. There have been

many studies on H-doped ZnO films^{4–8}, in which, a few of them focused on H and In co-doped ZnO⁶. In the report, however, the limitation is that the carrier concentration was very high, leading to low electron mobility (<30 cm²/Vs). Therefore, in this work, the combination of In and H in ZnO films is expected to obtain high conductivity owing to the moderate freeelectron amount (from In dopant), and good crystalline quality (high mobility due to H incorporation). We prepare successfully In and H co-doped ZnO (HIZO) thin films with low sheet resistance (R_S < 5 Ω /sq.), high electron mobility (> 40 cm²/Vs) and high average transmittance (>80%) in the wavelength range from 400 nm to 1100 nm.

MATERIALS & METHOD

The ceramic In-doped ZnO sputtering targets were synthesized by sintering ZnO and In_2O_3 powders at high temperature, which originated from Merck (Germany) and high purity (99.99%). The compositions of the targets were changed and listed in **Table 1**. The targets were used to deposit thin films on the glass substrate (Marienfeld, Germany) by using DC mag-

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 Table 1: The composition of ZnO sputtering targets

 with various In content

Targets	Α	В	С	D	E	F	G
Dopants	0.07	0.1	0.15	0.3	0.5	1.0	0.0
(at.% In)							

netron sputtering. For preparing HIZO thin films, a small amount of hydrogen gas (5N, SunAir, Singapore) was introduced into the sputtering atmosphere. The added amount of hydrogen was calculated through partial pressure ratio, $H_2/(H_2+Ar)$. The substrate temperature and sputtering power were maintained at 100°C and 60 W, respectively, while the target-substrate distance was fixed at 5 cm during the deposition process. At each In content in target, at least three thin films were deposited, so as to ensure repetition and accuracy in properties of the HIZO films.

The thickness of films was about 1000 nm, which was determined by using a Stylus profilometer (Veeco DEKTAK 6M, Korea). The carrier concentration, mobility, resistivity, and sheet resistance of the films were obtained from Hall measurement with Van der Pauw method (Ecopia HMS 3000, Korea) and the four-point probe. X-ray diffraction (BRUKER D8 Advance, US) was used to determine the crystalline structure of the films. The optical spectra were recorded by UV-Vis spectrophotometer (Jasco V-530, Japan) in the wavelength range of 300 - 1100 nm.

RESULT AND DISCUSSION

Electrical properties of the HIZO thin films from Hall measurement at room temperature are summarized in **Table 2**.

Through the electrical properties in **Table 2**, it is seen that the HIZO films sputtered from the B target at $H_2/(H_2+Ar) = 3.5\%$ obtain high electron mobility of $47.0 \text{ cm}^2/\text{Vs}$ and the lowest resistivity of 4.9×10^{-4} Ω .cm, which corresponds to the lowest sheet resistance of $4.7 \Omega/\text{sq}$. From these results, the combination of In and H in ZnO films initially proposes the significant enhancement in electrical properties of the HIZO films. To evaluate the simultaneous influence of In and H dopants, the optimum HIZO films are compared to the pure ZnO films and the IZO films (without H introduction). The electrical and optical parameters of the three films are listed in **Table 3**.

Table 3 shows that the carrier concentration of the sample B0 is higher than that of the sample G, but lower than that of the sample B. Slassi *et al.*⁹ and

Khuili et al.¹⁰ reported that when a Zn atom is substituted by a IIIA-group atom, the Al 3s, Ga 4s or 4p and In 5s orbitals contribute to the occupied states near the Fermi level, which acts as a donor state around the Fermi level. It may be considered as the origin of the increased carrier concentration and electrical conductivity of IIIA group-doped ZnO films. Furthermore, hydrogen also acts as a source contributing electrons for conduction, with shallow donor states below ~0.03 - 0.1 eV from the bottom of the conduction band ^{11,12}. The exciting thing is that the electron mobility of sample B reaches the highest values of 47.3 cm²/Vs. This value is considered much higher than that of the other study on HIZO films⁶. The reason can be from the excellent harmony of In and H dopants in the lattice structure of ZnO films. To demonstrate this hypothesis, the crystalline structure of the films are investigated through XRD spectra in Figure 1.

From Figure 1, it is seen that all the films only have a prominent ZnO (002) peak, indicating the typical hexagonal-wurtzite structure of ZnO material (JCPDS 36-1451). No crystalline phases of In compounds are observed in the X-ray patterns, suggesting that In³⁺ probably replaces Zn²⁺or locates in interstitial sites in ZnO lattices or segregates in the non-crystalline region at the grain boundaries¹³. The HIZO films (sample B) have the (002) peak with the highest intensity, which indicates that the addition of the small amount of In and H can give rise to significant improvement in the crystalline quality of ZnO films. Furthermore, the presence of hydrogen causing the shift of (002) peak in sample B towards lower 2θ angle as compared to sample G is observed. It suggests the reduction of defects in the crystalline structure of the sample B. Besides, the mean free paths (MFP) of electron in the sample G, sample B0 and sample B (2.5 nm, 3.2 nm and 7.1 nm, respectively) are much smaller than their crystal size (26.9 nm, 30.4 and 29.6 nm, respectively). Thus, the grain boundary scattering cannot be the dominant mechanism affecting the electron mobility. The increase in mobility value, as shown in Table 3, therefore, can be due to the decrease in ionized impurities scattering.

In literature, hydrogen can support In^{3+} substituting into Zn^{2+} sites due to charge neutrality. The replacement acts an essential role in increasing In^{3+} donors, which can be realized from the shift of (002) peak, as mentioned in XRD patterns. As a result, the reduction of scattering centers relating to interstitial impurities, which leads to increase the mobility. Furthermore, hydrogen can passivate some defects in the crystalline

Table 2: Carrier concentration (n), electron mobility (μ), resistivity (ρ) and sheet resistance (R _S) of the HIZO films							
Films deposited from targets	H ₂ /(H ₂ +Ar) (%)	n (10 ²⁰ cm ⁻³)	μ (cm²/Vs)	ρ (10 ⁻⁴ Ω.cm)	R _S (Ω/sq.)		
А		1.9	49.1	6.8	6.6		
В		2.7	47.3	4.9	4.7		
С		2.3	43.4	6.2	6.0		
D	3.5	2.6	36.1	6.7	6.5		
Е		3.9	21.7	7.3	7.2		
F		7.3	27.0	7.7	7.4		

Table 3: Carrier concentration (n), electron mobility (μ), resistivity (ρ), sheet resistance (R_S), and average transmittance in the visible (T_{Vis}) and the near IR regions of the ZnO (sample G), IZO (sample B0) and HIZO (sample B) films

Sam- ples	H ₂ /(H ₂ +Ar) (%)	n (x10 ²⁰ cm ⁻³)	μ(cm ² /Vs)	ρ (10 ⁻⁴ Ω.cm)	R _S (Ω/sq.)	T _{Vis} (%)	T _{NIR} (%)
G	0	0.7	30.1	28.3	27.8	80.5	82.8
B0	0	1.2	32.1	15.9	14.6	78.1	79.1
В	3.5	2.7	47.3	4.9	4.7	81.5	82.0



Figure 1: X-ray diffraction patterns (left) and variations in peak position and crystal size of the ZnO, IZO and HIZO films (right).

structure of ZnO, such as zinc vacancies (V_{Zn}), dangling bonds⁶⁻⁸. This hydrogen passivation can occur through the adsorption and bonding formation of O-H, Zn-H, or Zn-OH in crystalline grains, grain boundaries, and film's surface of ZnO films⁶.

During the deposition process, the effect of hydrogen on the electrical properties, especially on the mobility of the films can be observed. Another reason may be the etching phenomenon in hydrogen plasma producing excited hydrogen atoms¹⁴. These excited H atoms can make bonds with O atoms leading to the lack of O atoms, which increases the number of O vacancies and interstitial Zn. Therefore, the control of hydrogen pressure is also the most important factor deciding the electrical and structural properties of the HIZO thin films.

Figure 2 illustrates the surface morphology of the ZnO, IZO, and HIZO thin films. It is seen that the grain density of sample B seems to be highest, while the sample G has the lowest value. This is entirely consistent with the improvement in the crystalline structure and electrical properties of the films, as discussed in the XRD (**Figure 1**) and Hall measurement (**Tables 2 and 3**) results. Additionally, in sample B, the density of black spots tends to decrease. It is possible that In³⁺ ions can insert into the Zn vacancies.



Figure 2: FE-SEM images of the G, B0 and B samples.

Simultaneously, H^+ ions also fill up with O vacancies and the black spots are enlarged at the grain boundaries. This suggests that H^+ ions have been linked to O^{2-} ions at the surface, which removes small particles from the surface of thin films¹⁵.

As mentioned in **Table 3** and **Figure 3**, 1000-nmthick sample B has the lowest sheet resistance of 4.7 Ω /sq. and high average transmission over 80% in the Vis –NIR region, which can be well used as transparent electrodes for solar cells.

CONCLUSION

A small amount of 0.1 at.% In-mixed ZnO sputtering target and sputtering in hydrogen plasma are the optimum conditions for depositing good-performance ZnO thin films. The carrier concentration increases significantly from 0.7 to 2.7×10^{20} cm⁻³ due to In donors. The electron mobility enhances by 67%, thanks to the reasonable hydrogen ratio (3.5%). As a result, the sheet resistance also decreases by 83% from 27.8 to 4.7 Ω /sq. Through this work, we propose that the HIZO films can be used as transparent electrodes in low-temperature applications (100°C).

ABBREVIATIONS

DC: Direct Current

TCO: Transparent Conducting Oxides MFP: Mean Free Paths ITO: Sn-doped In_2O_3 XRD: X – Ray Diffraction V_{Zn} : Zinc Vacancy FE-SEM: Field Emission Scanning electron Microscopy Vis-NIR: Visible and Near Infrared Range

COMPETING INTERESTS

The authors declare no competing interests.

AUTHORS' CONTRIBUTIONS

Truong Huu Nguyen researched and wrote the paper. Vinh Cao Tran designed the study. Tinh Van Nguyen, Anh Tuan Thanh Pham, Dung Van Hoang, Hung Minh Vu, Hoi Cong Nguyen conducted the experiments. Bach Thang Phan help to revise the manuscript.

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Figure 3: Optical transmittance spectra of the G, B0 and B films.

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