

# Spray-Dried Encapsulation of a Roselle (*Hibiscus Sabdariffa*) and Evaluation of its Potential as a Rich-Antioxidant Compound

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## ABSTRACT

**Introduction:** There is a growing demand for the use of rich-antioxidant compounds from plant extracts to develop functional food products or biomedical products. In this work, we studied the impact of producing encapsulated roselle extracts with 1% w/v microencapsulated carriers from maltodextrin and xanthan gum as a powder rich in antioxidant properties. **Methods:** In particular, the roselle extract was subjected to spray drying using an acidified solution as a solvent. Combinations of maltodextrin and xanthan gum as wall materials at 1% w/v were first tested for the roselle. The characteristics of the spray-dried powder were confirmed by scanning electron microscopy (SEM), including its solubility, titratable acidity, anthocyanin content, and antibacterial and antioxidant activities. **Discussion:** The results showed that the addition of XTG directly affected the morphology and antioxidant activities of the encapsulated materials. The particles presented a spherical shape with diameters ranging from 1–5  $\mu\text{m}$ . The encapsulation of roselle extract via a specific combination of maltodextrin and xanthan gum (1% w/v) maintains the extract properties and controls its release properties. A reduction in antioxidant activity and solubility was observed with increasing concentrations of XTG. **Conclusion:** The findings of this study demonstrated that the obtained roselle spray-dried powders can be used as rich-antioxidant supplements for the food, biomedical, and cosmetic industries. The control of xanthan gum content leads to significantly slower release, which has implications for controlled delivery applications.

**Key words:** Roselle extract, microencapsulation, spray drying, antioxidant, microparticle

## 1 INTRODUCTION

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2 Naturally occurring compounds such as vitamins, polysaccharides, flavonoids, fatty acids, terpenes, and tannins serve as topical antioxidants by inhibiting oxidative stress<sup>1</sup>. The use of natural herbs in food and health has a rich history, often with minimal adverse side effects. Recent research highlights their potential applications in the food and beverage industry and regenerative medicine, emphasizing their biocompatibility and safety. Notably, functional compounds from plants, particularly those rich in phytochemicals such as anthocyanins, are gaining attention for their antioxidant properties<sup>2,3</sup>.

4 Roselle (*Hibiscus sabdariffa*), an edible plant, has garnered significant interest for its potential applications across various industries<sup>4</sup>. This shrub, which belongs to the *Malvaceae* family, is renowned in many countries as a folk medicinal plant. Roselles contain numerous beneficial chemical components, including anthocyanins, flavonoids, and polyphenols, which support cardiovascular health, prevent hypertension, reduce fever, and protect liver function. Additionally,

it is reported to possess antimicrobial, diuretic, digestive, and sedative properties. The anthocyanins in roselles, such as delphinidin 3-sambubioside, cyanidin 3-sambubioside, delphinidin 3-glucoside, and cyanidin 3-glucoside, are potent antioxidants, natural food colorants and antioxidant supplements for the cosmetic industry<sup>4</sup>. However, these compounds are relatively unstable and can degrade due to environmental factors. Therefore, the application of technology to the processing of natural active ingredients in hibiscus flowers has become a research focus to increase the value of this product in various applications.

5 Spray drying and encapsulation represent advanced techniques for preserving the stability and functionality of biological compounds after processing. This method involves protecting the active substance—whether solid, liquid, or gas—by encasing it within a polymer matrix or high-molecular-weight substrate<sup>5,6</sup>. This encapsulation process not only shields the active substance from environmental impacts but also maintains its properties and activity. Following encapsulation, the mixture undergoes spray drying to produce small particles of the active substances

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at the micro- or nanoscale. This process minimizes interactions with other environmental components, thereby preserving essential properties such as nutritional, color, antioxidant, and antibacterial characteristics. Despite the brief exposure to high temperatures during spray drying, this process can lead to the decomposition of compounds within the mixture<sup>5,7</sup>. Therefore, it is crucial to optimize factors such as spray drying temperature, drying conditions, and the composition of encapsulating agents to develop an effective drying process.

In this study, we aimed to produce roselle-derived powder as a natural colorant and antioxidant supplement. The primary variable was the ratio of encapsulating agent used in the spray-dried roselle powder. Maltodextrin and xanthan gum were employed as encapsulating agents at a concentration of 1% w/v. Typically, the use of encapsulating agents in spray drying typically requires approximately 10–30% w/v to optimize the spray dryer<sup>8–10</sup>. However, some studies have reported the possibility of producing microencapsulated powders with less than 10% encapsulating agent, even at concentrations lower than 3% w/v<sup>11,12</sup>. We hypothesize that reducing the amount of encapsulating agent may affect the efficacy of the spray dryer. However, reducing the concentration of encapsulating agent could result in a lower yield of solid powder but a higher proportion of extract content, benefiting the use and storage of the powder. Maltodextrin and xanthan gum are known to be compatible materials commonly applied in spray drying technology<sup>13</sup>. The presence of maltodextrin can improve drying and powder flowability, whereas the addition of xanthan gum, with its high colloidal ability, can enhance the encapsulated structure. As a thickening agent, xanthan gum is expected to support the encapsulation process of roselle extract via spray drying. After spray drying, the samples were dissolved, and their antioxidant capacity and anthocyanin content, which are key determinants of product applicability, were evaluated. Investigations were carried out on various characteristics, including the composition of the encapsulating carriers, the micromorphology of the roselle powder, and its physicochemical properties, to optimize the high antioxidant content of the spray-dried powder. Powder morphology was assessed by scanning electron microscopy. The anthocyanin content was measured on the basis of the cyanidin-3-O-sambubioside group, and the antioxidant activity was evaluated via the DPPH assay, which measures free radical scavenging ability. To our knowledge, this study is the first to introduce the use of maltodextrin and xanthan gum as encapsulating agents for roselle.

**MATERIALS AND METHODS**

**Materials**

Dry roselle (hibiscus) was purchased from Hichagol Co., Ltd., Hue, Vietnam. Acetic acid, citric acid, and sodium hydroxide were purchased from Xilong Chemical Co. Ltd. Potassium chloride and anhydrous sodium acetate were purchased from Guangdong Guanghua Sei-Tech Co. Ltd. Maltodextrin (DE: 10 - 12%) was received from Qinhuangdao Lihua Starch Co. Ltd. DPPH (2,2-diphenyl-1-picrylhydrazyl) and dimethyl sulfoxide (DMSO) were purchased from Sigma–Aldrich.

**Roselle extraction**

The dry roselle flower was ground into a fine powder with a particle size of approximately 0.5 cm. An acetic acid 2% extraction solvent was prepared for the extraction process. Roselle powder was immersed in the solvent at a ratio of 4 g of roselle powder to 100 ml of extraction solvent. The mixture was incubated at 4 °C. After 24 h, the extraction mixture was collected. The solid residue after filtration was then subjected to another extraction process for 24 h. The final extracted material was stored at 4 °C for characterization.

**Preparation of roselle microencapsulation**

Four different formulas were spray-dried. Before microencapsulation, each encapsulating agent was dispersed in the roselle extracted solution and mixed until homogenization was reached. The encapsulation concentration was 1% w/v. The components of the wall materials and sample design are presented in Table 1.

**Preparation of spray-dried roselle powder**

The SD-1000 Eyela (Eyela, Japan) was used to spray the encapsulating mixture. Once the mixtures were made, they were fed into the spray dryer. The inlet temperatures are set at 160 °C, and the outlet temperatures are set at 80 °C. A feed flow of the solution to be dried with a spray air flow of 0.7 m<sup>3</sup>/min and a pressure drop of 180 kPa was used. To maintain the homogeneity of the solution, the encapsulated solution was gently mixed via a pump. After spray drying, the powders were weighed and stored in closed plastic bags, after which they were stored at -20 °C until evaluation.

**Microencapsulation particle morphology**

The powders were sputter-coated with gold at 10 kV, and their microstructural characteristics were analyzed via SEM (JSM-IT100, JEOL, Japan). Different magnifications of the sample were recorded. The

149 size and morphological structure of the encapsulated  
150 roselle powder were measured via the ImageJ pro-  
151 gram, with 30 data points collected for each sample.

### 152 Solubility

153 The solubility of the powder was assessed via a  
154 method outlined in the literature<sup>14</sup>. In a concise pro-  
155 cedure, one gram of spray-dried powder was com-  
156 bined with 10 millilitres of distilled water, and the  
157 mixture was stirred via a magnetic stirrer for 5 min-  
158 utes. The mixture was subsequently subjected to cen-  
159 trifugation at 4000 revolutions per minute (rpm) for  
160 10 minutes. The resulting supernatant was carefully  
161 transferred to a Petri dish and subsequently dried at  
162 105 °C for 2 hours. The solubility was determined by  
163 calculating the difference in solid weight between the  
164 reconstituted solution and the original solid sample.

### 165 pH and total acidity

166 The total titrated acidity expressed as % citric acid  
167 was determined through acid titration, utilizing 0.1  
168 N NaOH and phenolphthalein as indicators. The ini-  
169 tial pH value of each sample was determined via a pH  
170 meter. Following the pH measurement, 2–3 drops of  
171 phenolphthalein indicator solution were introduced.  
172 The titration then commenced with the addition of 0.1  
173 N NaOH until the pH reached 7, signified by a color  
174 change in the solution. Throughout the titration pro-  
175 cess, the sample in the flask was swirled continuously.  
176 The volume of NaOH solution used in the titration  
177 was carefully noted. Finally, this recorded volume was  
178 used to calculate the total titrated acidity expressed as  
179 a percentage of citric acid. The equation for titratable  
180 acidity is as follows:

$$\% \text{ acid (titratable acidity)} = \frac{N \times V \times Eqwt}{W \times 1000} \times 100$$

181 where:

182 *N*: normality of titrant;

183 *V*: volume of titrant;

184 *Eqwt*: equivalent weight of predominant acid;

185 *W*: mass of sample.

### 186 Total anthocyanin content

187 The total anthocyanin content (TAC) of the spray-  
188 dried powder was determined following a method  
189 outlined in the literature with slight modifications<sup>15</sup>.  
190 In a concise procedure, 50 mg of the sample was ac-  
191 curately weighed and mixed with 1 ml of distilled wa-  
192 ter, achieving a homogenous solution through vor-  
193 tex mixing. Two buffer solutions, namely, 0.2 M KCl  
194 buffer (pH 1.0) and 0.1 M acetate buffer (pH 4.5),  
195 were employed to dilute the sample mixture. To assess

the absorbances of the diluted samples, measurements  
were taken at two distinct wavelengths, 520 nm and  
700 nm, via a microplate reader. The content (mg/g  
of dry matter) was calculated via the following equa-  
tion:

$$TAC = \frac{A \times M_w \times D_f \times 10^3}{\epsilon \times l} \quad (2)$$

$A = (A_{\lambda 520} - A_{\lambda 700}) \text{ pH } 1 - (A_{\lambda 520} - A_{\lambda 700}) \text{ pH } 4.5$

$D_f$ : volume of diluted samples;

$L$ : 96-well plate length (0.286 cm);

$M_w$ : molecular weight cyanidin-3-glucoside, 449.2  
g/mol;  $\epsilon = 26,900$  L/mol.

### DPPH free radical scavenging activity

DPPH, which stands for 2,2-diphenyl-1-  
picrylhydrazyl, serves as a reagent used to measure  
the antioxidant activity of substances. In brief, the  
procedure involves the preparation of spray-dried  
powders mixed with DI at different concentrations (5,  
10, 15, 30, and 45 mg/ml). DPPH was simultaneously  
dissolved in DMSO (0.1 mM). Subsequently, 20  
microliters of the sample at various concentrations  
were added to wells in a 96-well plate, along with  
180 microliters of the DPPH reagent. The reaction  
mixture was then incubated for 30 minutes in the  
dark, and the absorbance was measured at 517 nm.  
The percentage of DPPH-scavenging activity was  
calculated via the following equation:

$$I\% = \frac{A_0 - A_s}{A_0} \times 100\%$$

where:

$A_0$ : the optical density value of the blank;

$A_s$ : optical density value of the test sample.

### (1) Evaluation of the antibacterial activity of the roselle powder

The antibacterial activity of the roselle powder was  
evaluated via the use of two representative microor-  
ganisms: *Escherichia coli* (*E. coli*) (ATCC 25922), a  
gram-negative bacterium, and *Staphylococcus aureus*  
(*S. aureus*) (ATCC 25913), a gram-positive bacterium.  
The minimum inhibitory concentration (MIC) was  
determined via a classical dilution method adapted to  
96-well microtiter plates. Each assay was performed  
in triplicate. Serial dilutions of roselle powder were  
prepared in the wells, and each well was inoculated  
with a standardized bacterial suspension. The plates  
were then incubated under appropriate conditions for  
bacterial growth. The MIC value was defined as the  
lowest concentration of roselle powder that inhibited  
visible bacterial growth after the incubation period.

## 241 Statistical analysis

242 All the experiments were performed in triplicate, and  
243 the results are presented as the means  $\pm$  standard de-  
244 viations. Statistical analysis was conducted via one-  
245 way ANOVA for all tests, followed by Tukey's post hoc  
246 analysis to determine significant differences between  
247 groups. A p value of less than 0.05 was considered sta-  
248 tistically significant.

## 249 RESULTS

### 250 Physio-chemical profile of roselle extract

251 The extraction step for the dried roselle powder  
252 product was meticulously performed to isolate valu-  
253 able compounds such as anthocyanins, vitamins,  
254 flavonoids, and phenols from the roselle flower. In  
255 this study, cold brew extraction was performed with  
256 2% acetic acid utilized as the solvent, ensuring effi-  
257 cient extraction of these bioactive components. Ini-  
258 tially, the properties of the extract solution were eval-  
259 uated, with a focus on parameters such as pH, acid  
260 content, anthocyanins, and antioxidant capacity, be-  
261 fore proceeding to the spray drying stage.

262 During the initial evaluation, the solid content in the  
263 extracted sample was determined to be 10 mg/ml. The  
264 pH and acid content of the extract were measured  
265 via titration methods, revealing an acid content value  
266 of 2.8%. The anthocyanin content in the roselle ex-  
267 tract was  $50.06 \pm 2.19$  mg/L, indicating the signif-  
268 icant presence of these potent antioxidants. Addi-  
269 tionally, the antioxidant capacity, as assessed through  
270 DPPH scavenging activity, had an IC<sub>50</sub> value of  $2.83$   
271  $\pm 0.03$  mg/ml. The extraction process effectively eval-  
272 uated the anthocyanin, antioxidant, and total titrat-  
273 able acidity (TTA) contents of the initial roselle ex-  
274 tract. As anticipated, the roselle flowers presented  
275 high antioxidant capacity and high anthocyanin con-  
276 tent. Identifying these components at the extraction  
277 stage represents a crucial initial success in the over-  
278 arching goal of developing a roselle powder product  
279 with increased antioxidant and anthocyanin contents.  
280 The parameters of the extraction solution are summa-  
281 rized in Figure 1.

### 282 Microencapsulated-roselle powder by 283 spray drying

284 After the spray drying process, the morphology of the  
285 roselle powder was evaluated. In general, microen-  
286 capsulated roselle powder consists of fine particles in  
287 its dry form, despite some changes being observed  
288 among the coating and uncoating treatments. Fig-  
289 ure 2 shows images of the spray-dried roselle powder  
290 under different blending treatments. In its uncoated

291 form, the pure roselle powder exhibited a deep red  
292 color. Owing to the lack of coating, the use of pure  
293 extract in spray drying systems is challenging. The  
294 high-temperature drying treatment caused the sugar  
295 components in the crude roselle extract to reach the  
296 glass transition state and adhere to the drying sys-  
297 tem, resulting in only a minimal amount of pure pow-  
298 der remaining after the spray treatment. For partic-  
299 les coated with maltodextrin (MD) or xanthan gum  
300 (XG), at a 1% coating agent concentration, the roselle  
301 powder tends to be pinkish-red. The red shade is  
302 slightly lower than that of the uncoated powder be-  
303 cause of the presence of the coating agents.

304 The morphology and microscopic size of the roselle  
305 powder were also evaluated via SEM. Figure 3 shows  
306 microscopy images of the different blending formulas  
307 of the roselle powder. SEM analysis of the uncoated  
308 roselle powder revealed smooth surfaces with visible  
309 bridges between the particles, indicating agglomera-  
310 tion (Figure 3A). In contrast, powder formulations  
311 mixed with a coating agent have uniform round parti-  
312 cles of varying sizes depending on the content of the  
313 coating agent and the xanthan gum component. At a  
314 1% coating agent concentration, the particle size was  
315 approximately 3–4  $\mu$ m. Figure 3B shows that parti-  
316 cles coated with maltodextrin alone have smooth sur-  
317 faces with a 3.4  $\mu$ m diameter (Table 2). Conversely,  
318 particles coated with xanthan gum tend to have more  
319 wrinkled surfaces. This is because xanthan gum, a  
320 large molecular weight adhesive substance, forms dy-  
321 namic bonds with maltodextrin, creating high surface  
322 tension during the shaping process at the spray-dried  
323 nozzle tip. This increases the size of the spray-dried  
324 particles, but the structure rapidly shrinks when the  
325 temperature decreases, resulting in wrinkled particles  
326 with many grooves<sup>7,16,17</sup>, as shown in Figure 3. Owing  
327 to the relatively low coating content, the produced  
328 particles have a small diameter, corresponding to a  
329 large surface area ratio. In all SEM images of the sam-  
330 ples in the 1% sample group, particles formed. This  
331 confirms the success of using 1% w/v encapsulation  
332 agents to produce roselle microparticles. The addi-  
333 tion of xanthan gum effectively reduces the surface  
334 area of the particles, making the particles wink and  
335 corrupt compared with those coated with maltodex-  
336 trin alone. This phenomenon has also been observed  
337 in other works with gum-derived carriers<sup>5,18,19</sup>.

338 This increase in particle size is necessary to reduce  
339 the surface area ratio of the sample, thereby limiting  
340 the influence of temperature and environmental hu-  
341 midity on the extract molecules encapsulated in the  
342 carrier layer. This hypothesis can be confirmed by


Physio-chemical profile of Roselle extract		
Solid content (mg/ml)	10.6 ± 0.2	
pH	2.8	
TTA (%)	2.83 ± 0.03	
Anthocyanin (mg/L)	50.1 ± 2.2	
IC50 Antioxidant properties by DPPH (mg/ml)	2.8 ± 0.3	

Figure 1: Physio-chemical profile of roselle extract extracted with acetic acid.

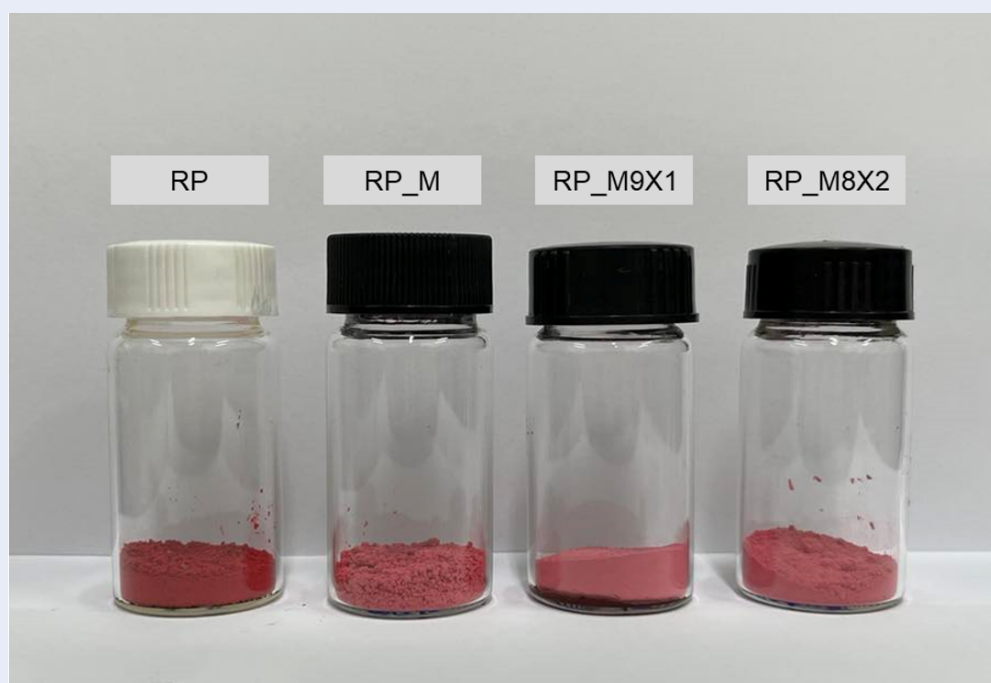


Figure 2: Spray-dried roselle powders subjected to various encapsulation treatments.

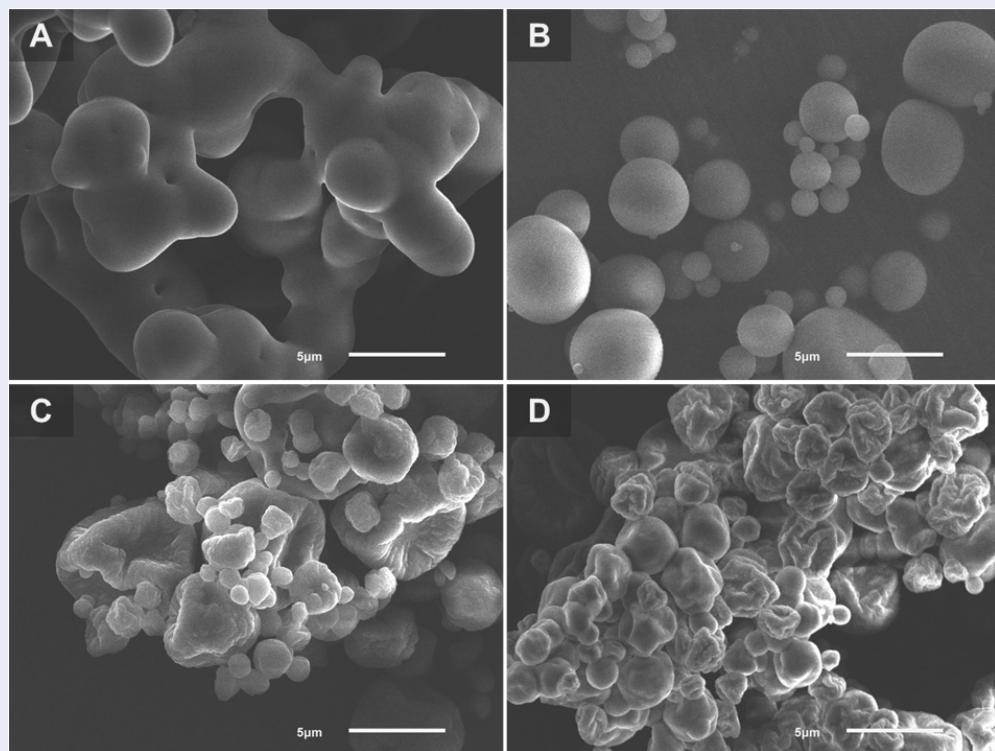
Table 1: Sample designed for microencapsulation of roselle extract.

Sample	Encapsulating material	
	MD (% w/v)	XG (% w/v)
RP	-	-
RP_M	1	
RP_M9X1	0.9	0.1
RP_M8X2	0.8	0.2

Table 2: Solubility and particle size of roselle spray-dried powders.

Sample	Solubility (%)	Particle size (µm)
RP_M	94.27 ± 0.28	2.6 ± 1.5
RP_M9X1	41.18 ± 1.38	2.7 ± 2.1
RP_M8X2	32.52 ± 1.54	2.8 ± 1.4





**Figure 3:** Micromorphology of roselle powder. (A) RP (roselle extract without encapsulation); (B) RP\_M; (C) RP\_M9X1; (D) RP\_M8X2. Scale bar: 5 μm.

**Table 3: Physio-chemical profile of the microencapsulated roselle powders.**

Sample	pH (2 mg/ml)	TTA (100 mg)	Anthocyanin	DPPH IC50 (mg/ml)	MIC	
					E. coli	S. aureus
RP	2.8 ± 0.1	32.9 ± 0.4 <sup>a</sup>	50.1 ± 2.2 <sup>a</sup>	2.8 ± 0.3 <sup>a</sup>	24.7 ± 0.9 <sup>a</sup>	31.8 ± 1.8 <sup>a</sup>
RP_M	2.4 ± 0.1	16.6 ± 1.3 <sup>b</sup>	41.3 ± 1.5 <sup>b</sup>	3.3 ± 0.1 <sup>a</sup>	17.5 ± 4.3 <sup>a</sup>	37.8 ± 1.5 <sup>a</sup>
RP_M9X1	2.9 ± 0.4	14.9 ± 0.7 <sup>b</sup>	38.4 ± 0.9 <sup>c</sup>	4.0 ± 0.5 <sup>a</sup>	41.3 ± 2.9 <sup>b</sup>	32.9 ± 0.4 <sup>a</sup>
RP_M8X2	2.4 ± 1.0	13.7 ± 2.0 <sup>b</sup>	38.0 ± 0.9 <sup>c</sup>	4.6 ± 1.1 <sup>a</sup>	35.1 ± 1.9 <sup>b</sup>	25.7 ± 3.1 <sup>a</sup>

Difference letters represent for significant different data at the level 0.05

343 the significant changes in the solubility of the mi-  
 344 croencapsulated material. As the content of XG in the  
 345 microencapsulated carrier increases, the solubility of  
 346 the powder decreases. At the highest content of XG  
 347 (RP\_M8X2), the solubility of the roselle powder was  
 348 only 32%, three times lower than that of the treatment  
 349 with only maltodextrin. This reduction shows that by  
 350 adding XG into the encapsulated component, the parti-  
 351 cles tend to be more hydrophobic, in which XG rein-  
 352 forces the microencapsulated wall, increasing the dif-  
 353 ficulty of penetration of water molecules into the parti-  
 354 cles.

**Physio-chemical profile of the microencap- 355  
 sulated roselle powder 356**

357 Table 3 presents the physicochemical profile of the  
 358 microencapsulated roselle powders. In general, the  
 359 characteristics of roselle powder differ among differ-  
 360 ent carrier treatments. The pH values of spray-dried  
 361 powder samples with varying concentrations of XG  
 362 and maltodextrin ranged from 2.4 to 2.9. The total  
 363 acidity values for RP\_M, RP\_M9X1, and RP\_M8X2  
 364 are between 13.7% and 16.6%, showing minimal vari-  
 365 ation among formulations. This consistency sug-  
 366 gests that the XTG concentration does not signifi-

cantly impact overall acidity. Hibiscus extract, which contains anthocyanins, is used in the production of these powders. Anthocyanins, which are responsive to pH variations, exist in four forms: quinonoidal bases, flavylium cations, carbinol (pseudobase), and chalcones<sup>20</sup>. At low pH, anthocyanins are predominantly in the red flavylium cation form. As the pH increases (>5), they shift to the blue/violet quinonoidal base or colorless carbinol/chalcone forms. pH values below 5 indicate an acidic environment, preserving the red color of the anthocyanins. The total anthocyanin content in the samples obtained with maltodextrin and xanthan gum as carrier agents was  $41.3 \pm 1.5$ ,  $38.4 \pm 0.9$ , and  $38.0 \pm 0.9$  mg/g for RP\_M, RP\_M9X1, and RP\_M8X2, respectively. The similar anthocyanin contents across formulations suggest that the XTG concentration does not significantly affect anthocyanin levels. Maltodextrin and xanthan gum provide a protective coating, enhancing anthocyanin stability by reducing susceptibility to nucleophilic attacks by water molecules. This protective mechanism is crucial for maintaining anthocyanin stability and efficacy in various applications, ensuring uniformity despite variations in carrier agents<sup>21</sup>. The IC<sub>50</sub> values of spray-dried powder samples with maltodextrin and xanthan gum as carrier agents were  $3.3 \pm 0.1$ ,  $4.0 \pm 0.5$ , and  $4.6 \pm 1.1$  mg/ml for RP\_M, RP\_M9X1, and RP\_M8X2, respectively. A higher xanthan gum content in the encapsulated matrix reduced the antioxidant activity of the powder, likely because the colloidal properties of xanthan gum limit the release of the core extract. This trend was also observed in the MIC results for the gram-negative *E. coli* strain. Detailed data on bacterial growth inhibition are presented in Table 3.

## DISCUSSION

To investigate the production of hibiscus powder with high antioxidant capacity, a spray drying method using maltodextrin and xanthan gum was employed in this study. To perform the spray drying method effectively, this study involved combining an effective extraction process with potential encapsulation treatments at 1% w/v to create roselle spray-dried powder. Our results confirmed the successful creation of hibiscus powder from the extract. Spray drying experiments using maltodextrin and maltodextrin mixed with xanthan gum produced hibiscus powder at various concentrations and encapsulation ratios. On the basis of the results presented in Table 3, the pH, TTA, anthocyanin content, antioxidant activity, and MIC of the roselle powder varied across the different experimental groups, both coated and uncoated. Notably, the uncoated powders presented higher indices

than their coated counterparts did. When the roselle powder was mixed with a 1% coating agent, the solid content ratio of the extract to the coating agent was 1:1. Consequently, although the 1% powder group had an anthocyanin content of 41 mg/g, the actual extract powder content in the dry product was effectively 80 mg/g, demonstrating the efficacy of the encapsulator in protecting the extract components during spray drying. Studies on the antibacterial effects of polyphenols indicate that their antibacterial activity can be derived from these anthocyanins, phenolic compounds, and flavonoids from the plant extract<sup>22-24</sup>. These extracts also contain anthocyanins such as delphinidin 3-sambubioside, delphinidin 3-glucoside, and cyanidin 3-sambubioside, which are responsible for their antioxidant and antibacterial properties<sup>23</sup>. Flavonoids, acids, and aromatic esters in the extract act on the cell walls of microorganisms. The encapsulated roselle powder inhibited *S. aureus* at 25–37 mg/ml and *E. coli* at 17–40 mg/ml. This finding is consistent with other studies showing weaker antibacterial activity of roselle extracts against gram-positive bacteria, possibly due to the thicker peptidoglycan membrane structure of these bacteria. According to Lim et al.<sup>25</sup>, the potential antibacterial activity of roselle against harmful microorganisms in foods has been demonstrated<sup>25</sup>. Owing to its antibacterial properties, the use of roselle powder has potential in food, meat products, and juices because of its ability to prevent bacterial growth and prolong shelf-life. Among the encapsulation treatments, the sample containing maltodextrin presented the highest anthocyanin content. Xanthan gum, owing to its large molecular weight, forms a more wrinkle encapsulation structure when blended with maltodextrin<sup>26</sup>. However, this structure affects the release process of the encapsulated components. This occurred because of the solubility of the powder, as the higher the amount of XG in the encapsulation component was, the lower the solubility was. Unlike previous studies that focused on single encapsulating agents, this research investigated the synergistic effect of maltodextrin and xanthan gum on the encapsulation efficiency and release profile of roselle extract. When xanthan gum was added, the anthocyanin content and antioxidant capacity, which were determined on the basis of the dry weight of the product, decreased slightly. These findings indicate that the 10% (RP\_M9X1) and 20% (RP\_M8X2) xanthan gum ratios affect the release of anthocyanin in the encapsulated matrix. As a higher xanthan gum content reduced the solubility of the powder product, these findings highlight the potential applications of

473 roselle powder coated with xanthan gum and mal-  
 474 todextrin. Moreover, roselle powder, with its high  
 475 antioxidant capacity and anthocyanin content, can be  
 476 used for various purposes, such as food coloring, an-  
 477 tioxidant supplements, or ingredients in drug delivery  
 478 systems with high antioxidant properties<sup>1,27,28</sup>. Our  
 479 results revealed a novel relationship between the xan-  
 480 than gum concentration and the release rate of an-  
 481 thocyanins, with higher xanthan gum content leading  
 482 to significantly slower release, which has implications  
 483 for controlled delivery applications. The lower solu-  
 484 bility of the RP\_M9X1 and RP\_M8X2 powders allows  
 485 for slower release, which is beneficial for prolonged  
 486 effects. In contrast, RP\_M powder, with its good wa-  
 487 ter solubility, is suitable for applications as an indus-  
 488 trial colorant or a soluble food product with high an-  
 489 tioxidant content. In conclusion, the 1% coating com-  
 490 position of maltodextrin and xanthan gum is suitable  
 491 for producing spray-dried roselle powder, effectively  
 492 protecting the extract's properties during the drying  
 493 process. In addition, the solubility of the release  
 494 rate of the microencapsulated powder can be con-  
 495 trolled by the presence of xanthan gum in the coating.  
 496 These experiments demonstrate the feasibility of us-  
 497 ing this coating system of maltodextrin and xanthan  
 498 gum at 1% w/v in spray drying to produce an effective  
 499 roselle powder. Overall, the results of the antioxidant  
 500 properties, antibacterial activity, and physio-chemical  
 501 characteristics of hibiscus powder highlight its poten-  
 502 tial for various applications.

503 **CONCLUSIONS**

504 In this study, spray drying experiments using mal-  
 505 todextrin and maltodextrin mixed with xanthan gum  
 506 successfully produced hibiscus powder at various  
 507 concentrations and encapsulation ratios. A 1% w/v  
 508 encapsulation mixture effectively formed small spray-  
 509 dried granules (1–5 μm). Hibiscus powder with-  
 510 out encapsulation was unsuitable for spray drying  
 511 because of the difficulty in obtaining powder sam-  
 512 ples. Treatment with only maltodextrin resulted in  
 513 stable values with relatively high anthocyanin con-  
 514 tents, and antioxidant properties were observed. With  
 515 a 1% w/v coating agent, an increased xanthan gum  
 516 content decreased the stability of the roselle powder,  
 517 with the lowest anthocyanin and DPPH values in the  
 518 RP\_M8X2 sample. SEM images revealed more wrin-  
 519 kled and distorted particles with xanthan gum. The  
 520 presence of XG reduces the surface area and may re-  
 521 inforce the coating agent composition, which can en-  
 522 hance protection from environmental influences. De-  
 523 spite these findings, all the experiments revealed that  
 524 spray-dried roselle seeds retained their antioxidant

capacity and color, as indicated by the anthocyanin 525  
 and IC50 results from the DPPH experiments. These 526  
 findings confirm the successful production of roselle 527  
 powder with high antioxidant capacity, suggesting 528  
 significant potential applications in food, cosmetics, 529  
 and pharmaceuticals for regenerative medicine. 530

**LIST OF ABBREVIATIONS USED** 531

- DPPH: 2,2-diphenyl-1-picrylhydrazyl 532
- DMSO: Dimethyl sulfoxide 533
- SEM: scanning electron microscopy 534
- TAC: total anthocyanin content 535
- MIC: minimum inhibitory concentration 536
- TTA: total titratable acidity 537
- MD: maltodextrin 538
- XG: xanthan gum 539

**COMPETING INTERESTS** 540

The authors declare that they have no competing in- 541  
 terests. 542

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**AUTHORS' CONTRIBUTION** 550

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